

## Alerts

## Charts and Tables

## Appliance Status

Appliance Name	Performance Alerts peaks, last 7 days	Requests per second	McAfee Anti-Malware Versions					URL Filter	
			Last update	Gateway Engine	Gateway DATs	Engine	DATs	Last update	version
lab-gateway			8 21 minutes ago	7501.1302.1844	1761	1750	7612	21 minutes ago	1.0.77

## Alerts

## Reset Filters

Appliance Filter	Event Filter	Message Filter	View to filter alerts
All	All	All	0
lab-gateway	06-Jul-2015 08:55:25:007	Login successful for user "james@lab" (191.168.1.10) (Origin: mgw-a, ID: 1760)	
lab-gateway	06-Jul-2015 08:58:12:007	User "admin" saved changes Successful (Origin: mgw-a, ID: 1761, 12 times within last 25 minutes)	
lab-gateway	06-Jul-2015 08:53:02:007	248 OUs have been updated (Origin: Certificate Chain Filter, ID: 1680)	
lab-gateway	06-Jul-2015 08:48:24:007	Connected to 1 server(s) in domain <2> (Origin: Authentication, ID: 961, 2 times within last 8 minutes)	
lab-gateway	06-Jul-2015 08:44:04:007	Login successful for user "admin" (191.168.1.100) (Origin: mgw-a, ID: 1756, 2 times within last 68 seconds)	
lab-gateway	06-Jul-2015 08:41:18:007	System info update successful (Origin: Updater, ID: 104)	
lab-gateway	06-Jul-2015 08:41:18:007	All OUs are up to date (Origin: Authentication Filter, ID: 943)	
lab-gateway	06-Jul-2015 08:41:03:007	Auto-update was successful. New version identifier is "Auto (Engine=1.4.3.18.18   Auto (DP=1.1.240.42   Auto (Setup=1.4.6.10" (Origin: Auto, ID: 159)	
lab-gateway	06-Jul-2015 08:38:32:007	3 minute load average exceeds selected limit (4.56 / 3.00). (Origin: Health monitor, ID: 24, 2 times within last 11 seconds)	
lab-gateway	06-Jul-2015 08:34:58:007	McAfee Gateway Anti-Malware update was successful. New version identifier is "AM (DAT=1761   AM (Engine=7501.1302.1844   MPF (DAT=7612   MPF (Engine=1750" (Origin: McAfee Gateway Anti-Malware, ID: 150)	
lab-gateway	06-Jul-2015 08:34:37:007	GTI web database update was successful. New version identifier is "GTI (Engine=1.4.3.18.18   GTI (Database=0.0.0.0" (Origin: GTI web database, ID: 119)	
lab-gateway	06-Jul-2015 08:34:07:007	The DLP Filter update was successful. New version identifier is "DLP (Engine=1.4.6.101   DLP (Agent=1.4.6.10" (Origin: McAfee Web Gateway (DLP Filter, ID: 140)	
lab-gateway	06-Jul-2015 08:34:07:007	The application control database (apptron-ic) update was successful. New version identifier is "Apptron (DB=4.1.0" (Origin: McAfee Gateway Application Control, ID: 160)	
lab-gateway	06-Jul-2015 08:29:57:007	Cloud Firewall Filter update was successful. New version identifier is "FW (DAT=1.4.6.10" (Origin: McAfee Web Gateway (Cloud Firewall Filter, ID: 140)	

# Mcafee Web Gateway Manual

**Rosina Ehmann**



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web where the salt is the conjugate base of the acid the four parts of the titration curve are described below and you should look to the appropriate text section to see how they are treated pure acid 0 mL of base is added section 17.3.2.1

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web solution a titrant volume 0 mL the solution pH is due to the acid ionization of  $HCl$  because this is a strong acid the ionization is complete and the hydronium ion molarity is 0.100 M the pH of the solution is then  $pH = -\log 0.100 = 1.000$  pH log 0.100 = 1.000 b titrant volume 12.50 mL

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web expert answer 100 2 ratings the example is based upon henderson hasselbalch equation the equation gives relation between ph and pha of weak acid the equation is as  $\text{ph} = \text{pka} + \log \frac{\text{salt}}{\text{acid}}$  at equivalence point all a view the full answer transcribed image text

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web unlike strong acids or bases the shape of the titration curve for a weak acid or base depends on the  $\text{pk a}$  or  $\text{pk b}$  of the weak acid or base being titrated a solution ph as a function of the volume of 1 00 m naoh added to 10 00 ml of 1 00 m solutions of weak acids with the indicated  $\text{pk a}$  values

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web titration curves source openstax chemistry atoms first 2e strong acid strong base titrations when titrating a strong acid such as hydrochloric acid with sodium hydroxide you are reacting the hcl directly with naoh  $\text{hcl aq} + \text{naoh aq} \rightarrow \text{h}_2\text{o l} + \text{nacl aq}$  hcl a q naoh a q h 2 o l nacl a q

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web ab acid base titration curves purpose the purpose of this lab is to determine the shape of acid base titration curves theoretically and experimentally and to use an acid base titration to measure the effectiveness of antacid tablets summary a titration curve is the plot of the ph value of a solution vs the amount of titrant added

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web aug 30 2022 acid base titrations are usually used to find the amount of a known acidic or basic substance through acid base reactions the analyte titrand is the solution with an unknown molarity the reagent titrant is the solution with a known molarity that will react with the analyte

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web acid base reactions can be used to determine concen tr ations of an acid or a base using stoic hiometric calculations and

titrimetric analysis the volume of a solution required to react

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web background plot of the pH of a solution against the volume of titrant added is called a titration curve the pH can be measured directly with a pH meter while titrant is added from a buret for the acids used in this experiment the titrant will always be a 0.1 M solution of the strong base NaOH

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web strong acids and strong bases completely ionize in solution resulting in water and a salt for a weak acid there is only partial ionization the titration curve of a strong base strong acid showed a stretched out curve as it started with a slow gradual change in pH as it reached the equivalence point

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