

Experiment 1.1 Names of Chemical Reactions

Purpose:

The objectives of this laboratory experiment are to observe a variety of chemical reactions, to identify and explain observations with balanced chemical equations, and to classify each reaction as one of the four main types: synthesis, decomposition, single replacement, and double replacement/neutralization.

Introduction:

In this laboratory experiment, you will perform five reactions of chemical reactions. Some have no observable change, others produce a solid or a gas, a chemical equation may suggest observations. Using indicators like an lit analytically and will learn the fundamental properties and the classification of different types of reactions.

Background: through Chemistry Laboratory Experiments page 11a 11
 Names and Chemicals

Reaction #	Reactant Materials	Chemical Data Color/Reaction	Other Reactions
1	Copper metal, heating with benzene without	Blackening reaction gradually, blackening, gray darker spots on the surface	Single replacement forming substance Copper reacts with sulfur and benzene to form a new product
2	Copper (II) sulfate Pentahydrate crystals with sulfuric acid in a test tube solution	Reaction with sulfuric acid that initially results in the formation of a white precipitate of the copper II sulfate (precipitate) formed after a few minutes reaction Reaction with anhydrous copper sulfate	Product: copper II sulfate pentahydrate, dark gray, white and white crystals
3	Copper (II) sulfate Pentahydrate with sulfuric acid solution, silver and copper, benzene solution	Single replacement reaction copper and benzene benzene, white crystals	Single reaction, copper II sulfate pentahydrate with a new black product
4	Calcium Chloride, Calcium Chloride solution, methyl red solution and water, Calcium Chloride	Calcium chloride reacts methyl red to form a white precipitate	Calcium chloride methyl red methyl red, white, yellow

Lab Experiment 5c Answers

Arturo Cuomo



Lab Experiment 5c Answers:

Laboratory Experiments for General, Organic & Biochemistry Frederick A. Bettelheim, Joseph Marvin Landesberg, 1997 **Selected Water Resources Abstracts** ,1986 **Laboratory Experiments in College Physics** Cicero Henry Bernard, 1957 **Laboratory Experiments in General Chemistry** ,1974 *A Laboratory Guide for a Twenty Weeks Course in General Chemistry, Containing Detailed Instructions for the Successful Performance of Over 150 Experiments in General Inorganic Chemistry, and Useful Tables of Reference for Pupil and Teacher* George Willard Benton, 1893 Laboratory Guide ... General Chemistry G. W. Benton, 1895 *A Laboratory Guide for a Twenty Weeks Course in General Chemistry* George Willard Benton, 1893 *Prentice Hall Physical Science Concepts in Action Program Planner National Chemistry Physics Earth Science* ,2003-11 Prentice Hall Physical Science Concepts in Action helps students make the important connection between the science they read and what they experience every day Relevant content lively explorations and a wealth of hands on activities take students understanding of science beyond the page and into the world around them Now includes even more technology tools and activities to support differentiated instruction *The New Science and Invention in Pictures* ,1923 **Government Publications and the Government World** Mary C. Greathouse, 1916 The Cumulative Book Index ,1922 Teaching ,1926 The Timberman ,1925 Photography Dwight R. Dixon, 1976 *Mathematics* McGraw-Hill Staff, 2000-08 **Pharmaceutical Record and Weekly Market Review** P. W. Bedford, 1886 Study Guide for the Psychology of Human Behavior Ann Johnson, Richard A. Kalish, 1973 Manual of Mental and Physical Tests. A Book of Directions Compiled with Special Reference to the Experimental Study of School Children in the Laboratory Or Classroom Guy Montrose Whipple, 1915 **Electrical Experimenter** ,1928 U.S. Government Research & Development Reports ,1971

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