

# Student Example Lab Report

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## **Title:**

Intermolecular Strengths of Substances

## **Objective(s):**

- To compare the boiling points of three substances
- To infer relative intermolecular strengths of substances using their boiling points

## **Hypothesis:**

I predict that vinegar has stronger intermolecular forces, and therefore a higher boiling point, compared to apple juice.

## **Procedure:**

### Materials: (List all the items used)

- Two safe substances to boil (apple juice, orange juice, milk, vinegar, cooking wine, saltwater)
- Heat source (stove or hot plate)
- Measuring cup
- Thermometer
- Container for heating (pot or beaker)

### Variables: (List the controlled, independent, and dependent variables of the lab.)

1. The controlled variables are the heat source, thermometer, container type, volume of liquid, and the boiling point of water (100 °C or 212 °F).
2. The independent variable (the variable we are changing) is the type of liquid heated.
3. The dependent variable (the variable that changes due to the independent variable) is the boiling point of each liquid.

### Summary of Steps:

1. Select two safe substances to boil. Make sure to list the units used to measure your substance when recording your volumes in the data chart.
2. Measure ½ to a full cup of your first substance and pour it into your heating container.

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**Yicheng Fang**



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## **Table of Contents Lab Manual Report Science Student Successful**

1. Understanding the eBook Lab Manual Report Science Student Successful
  - The Rise of Digital Reading Lab Manual Report Science Student Successful
  - Advantages of eBooks Over Traditional Books
2. Identifying Lab Manual Report Science Student Successful
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Lab Manual Report Science Student Successful
  - User-Friendly Interface
4. Exploring eBook Recommendations from Lab Manual Report Science Student Successful
  - Personalized Recommendations
  - Lab Manual Report Science Student Successful User Reviews and Ratings
  - Lab Manual Report Science Student Successful and Bestseller Lists
5. Accessing Lab Manual Report Science Student Successful Free and Paid eBooks
  - Lab Manual Report Science Student Successful Public Domain eBooks
  - Lab Manual Report Science Student Successful eBook Subscription Services
  - Lab Manual Report Science Student Successful Budget-Friendly Options

6. Navigating Lab Manual Report Science Student Successful eBook Formats
  - ePub, PDF, MOBI, and More
  - Lab Manual Report Science Student Successful Compatibility with Devices
  - Lab Manual Report Science Student Successful Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Lab Manual Report Science Student Successful
  - Highlighting and Note-Taking Lab Manual Report Science Student Successful
  - Interactive Elements Lab Manual Report Science Student Successful
8. Staying Engaged with Lab Manual Report Science Student Successful
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Lab Manual Report Science Student Successful
9. Balancing eBooks and Physical Books Lab Manual Report Science Student Successful
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Lab Manual Report Science Student Successful
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Lab Manual Report Science Student Successful
  - Setting Reading Goals Lab Manual Report Science Student Successful
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Lab Manual Report Science Student Successful
  - Fact-Checking eBook Content of Lab Manual Report Science Student Successful
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
  - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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