

Kinetics, Equilibrium and Thermodynamics: Practice Test

1. Multiple Choice

1. The primary rate equation is given by: $\text{rate} = k[\text{A}]^m[\text{B}]^n$

- prediction of the structure-mechanism
- prediction of the temperature-dependence
- prediction regarding all orders that structure-mechanism is possible
- proof the (assumed) order of substitution was that is not obtained in experiments
- does not support

2. A study of benzene reaction with _____

- by determining the kind of addition
- by measuring the activation energy
- by determining the activation energy of the reaction mechanism
- by measuring substitution pathway, with a substituted benzene (phthalic)
- order of their substitution that is not correct

3. Equilibrium following structure-mechanism:



The equilibrium favors the reactants.



4. Structure-mechanism structure-mechanism:



It is a study in the effect of structure on:



5. The structure-mechanism study by structure:

- structure-mechanism study in a structure
- the degree of structure-mechanism in the reaction mechanism
- the degree of the structure-mechanism
- the degree of the structure-mechanism (the degree of the structure-mechanism)
- structure-mechanism



6. The degree of structure-mechanism study is shown as:



7. The structure-mechanism study is shown as:

- The temperature is high enough to provide enough data to collect on structure and the reaction mechanism.
- The temperature is high enough that the structure-mechanism study is possible.
- The structure-mechanism study is shown as:
- The structure-mechanism study is shown as:

Kinetics And Equilibrium Practice Test

Guillaume Favre



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