

Isotopes and average atomic mass worksheet

Determine the weighted average atomic mass of the following elements.

Element	Isotopes	Average Atomic Mass
Iodine (I)	a) 17% ^{126}I b) 80% ^{127}I c) 3% ^{128}I	
Gold (Au)	a) 50% ^{197}Au b) 50% ^{198}Au	
Carbon (C)	a) 98% ^{12}C b) 2% ^{14}C	
Iron (Fe)	a) 15% ^{55}Fe b) 85% ^{56}Fe	
Hydrogen (H)	a) 99% ^1H b) 0.8% ^2H c) 0.2% ^3H	
Nitrogen (N)	a) 95% ^{14}N b) 3% ^{15}N c) 2% ^{16}N	
Sulfur (S)	a) 95% ^{32}S b) 4.22% ^{34}S c) 0.76% ^{33}S d) 0.014% ^{36}S	
Chlorine (Cl)	a) 75.53% ^{35}Cl b) 24.47% ^{37}Cl	

Isotopes And Average Atomic Mass Answer Key

CH Cherryholmes



Isotopes And Average Atomic Mass Answer Key:

Reviewing **Isotopes And Average Atomic Mass Answer Key**: Unlocking the Spellbinding Force of Linguistics

In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is truly astonishing. Within the pages of "**Isotopes And Average Atomic Mass Answer Key**," an enthralling opus penned by a highly acclaimed wordsmith, readers attempt an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve in to the book is central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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