

- Directions:**
 1. An element may be assigned a positive or negative charge depending on whether it is a metal or nonmetal.
 2. An atom with a positive charge is a cation.
 3. An atom with a negative charge is an anion.
 4. The number of protons in an atom is the atomic number.

Ion Practice Set

1. Write the symbol for the ion that has the following number of protons and electrons.
 2. Write the symbol for the ion that has the following number of protons and electrons.
 3. Write the symbol for the ion that has the following number of protons and electrons.

Complete the following table, using the periodic table in the back of your book.

	Element Name	Ion Symbol	Number of Protons	Number of Electrons	Charge on Ion (lost or gained)	Ion Name
1	Sodium	Na^+	11	10	lost one	cation
2	Chlorine	Cl^-	17	18	gained one	anion
3	Calcium	Ca^{2+}	20	18	lost two	cation
4	Oxygen	O^{2-}	8	10	gained two	anion
5	Aluminum	Al^{3+}	13	10	lost three	cation
6	Sulfur	S^{2-}	16	18	gained two	anion
7	Hydrogen	H^+	1	0	lost one	cation
8	Lithium	Li^+	3	2	lost one	cation
9	Fluorine	F^-	9	10	gained one	anion
10	Magnesium	Mg^{2+}	12	10	lost two	cation
11	Strontium	Sr^{2+}	38	36	lost two	cation
12	Barium	Ba^{2+}	56	54	lost two	cation
13	Iron	Fe^{2+}	26	24	lost two	cation
14	Copper	Cu^+	29	28	lost one	cation
15	Gold	Au^+	79	78	lost one	cation
16	Mercury	Hg^{2+}	80	78	lost two	cation
17	Lead	Pb^{2+}	82	80	lost two	cation
18	Platinum	Pt^{2+}	78	76	lost two	cation
19	Silver	Ag^+	47	46	lost one	cation
20	Vanadium	V^{3+}	23	20	lost three	cation
21	Chromium	Cr^{3+}	24	21	lost three	cation
22	Manganese	Mn^{2+}	25	23	lost two	cation
23	Zinc	Zn^{2+}	30	28	lost two	cation
24	Cadmium	Cd^{2+}	48	46	lost two	cation
25	Barium	Ba^{2+}	56	54	lost two	cation
26	Strontium	Sr^{2+}	38	36	lost two	cation
27	Calcium	Ca^{2+}	20	18	lost two	cation

Ion Practice Set Answer Key

Didier Musso



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