Name \_\_\_\_\_ Date \_\_\_\_ Period \_\_\_\_

### **NOMENCLATURE PACKET**

#### Worksheet I: Binary Ionic Compounds (representative metals)

- 1. Name the following ionic compounds:
  - a. Al<sub>2</sub>O<sub>3</sub> Aluminum oxide
  - b. Cs<sub>2</sub>O Cesium oxide
  - c. Rb<sub>3</sub>N Rubidium nitride
  - d. CasNe Calcium nitride
  - e. SrSe Strontium selenide
  - f. Cs:S Cesium sulfide
  - g. Al<sub>2</sub>S<sub>2</sub> Aluminum sulfide
  - h. LiBr Lithium bromide
  - i. MgsNs Magnesium nitride
  - j. CaF<sub>2</sub> Calcium fluoride
- 2. Write the chemical formula for the following ionic compounds:
  - a. Barium nitride BarNe
  - b. Indium Fluoride InF
  - c. Calcium oxide CaO
  - d. Sodium nitride Na<sub>3</sub>N
  - e. Magnesium chloride MgClz
  - f. Potassium oxide KO
  - g. Magnesium oxide MgO
  - h. Potassium sulfide K25
  - i. Lithium nitride LisN
  - j. Strontium fluoride SrFz
  - k. Aluminum sulfide Al<sub>2</sub>S<sub>3</sub>
  - I. Calcium oxide CaO

# **Ionic Nomenclature Review Vocabulary Answer Key**

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