

CHARGES, NAMES, AND FORMULAS OF COMMON IONS

Positive Ions (Cations)

+1 Charge

Name	Formula
Ammonium	NH_4^+
Copper(I)	Cu^+
Lithium	Li^+
Potassium	K^+
Silver	Ag^+
Sodium	Na^+

+2 Charge

Name	Formula
Barium	Ba^{2+}
Calcium	Ca^{2+}
Copper(II)	Cu^{2+}
Iron(II)	Fe^{2+}
Lead(II)	Pb^{2+}
Magnesium	Mg^{2+}
Nickel(II)	Ni^{2+}
Strontium	Sr^{2+}
Tin(II)	Sn^{2+}
Zinc	Zn^{2+}

+3 Charge

Name	Formula
Aluminum	Al^{3+}
Iron(III)	Fe^{3+}

+4 Charge

Name	Formula
Lead(IV)	Pb^{4+}
Tin(IV)	Sn^{4+}

Negative Ions (Anions)

-1 Charge

Name	Formula
Acetate	$\text{C}_2\text{H}_3\text{O}_2^-$
Bromide	Br^-
Chloride	Cl^-
Dihydrogen Phosphate	H_2PO_4^-
Fluoride	F^-
Hydroxide	OH^-
Hydrogen Carbonate (bicarbonate)	HCO_3^-
Hydrogen Sulfate (bisulfate)	HSO_4^-
Hypochlorite	ClO^-
Iodide	I^-
Nitrate	NO_3^-
Permanganate	MnO_4^-
Thiocyanate	SCN^-

-2 Charge

Name	Formula
Carbonate	CO_3^{2-}
Chromate	CrO_4^{2-}
Dichromate	$\text{Cr}_2\text{O}_7^{2-}$
Hydrogen Phosphate	HPO_4^{2-}
Oxide	O^{2-}
Oxalate	$\text{C}_2\text{O}_4^{2-}$
Sulfate	SO_4^{2-}
Sulfite	SO_3^{2-}

-3 Charge

Name	Formula
Nitride	N^{3-}
Phosphate	PO_4^{3-}

Ions And Their Chemical Formula Answers

John T. Moore, Richard H. Langley



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Pharmaceutical Journal** ,1911 **The Ohio Teacher** ,1906 *The Journal of Industrial and Engineering Chemistry*
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Physical Chemistry with Practical Applications** Edmund Brydges Rudhall Prideaux,1912

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In an electronic era where connections and knowledge reign supreme, the enchanting power of language has become much more apparent than ever. Its power to stir emotions, provoke thought, and instigate transformation is really remarkable. This extraordinary book, aptly titled "**Ions And Their Chemical Formula Answers**," published by a highly acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound effect on our existence. Throughout this critique, we shall delve to the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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