

Final Study Guide

Buoyancy - ability of a fluid to exert an upward force on object immersed in a fluid. To float, an object must displace an amount of water that has a mass greater than the mass of the object. This usually happens in things that are hollow or filled with air.

Viscosity - a fluid's resistance to flow or how thick, or hard to pour a fluid is.

Density - mass per unit volume. Density = $\frac{\text{mass}}{\text{volume}}$

Physical Property - a property that can be measured without affecting the substance.
Ex: color, shape, size, mass, anything that can be observed with your senses

Chemical Property - a property that tells us how the substance interacts with other substances, the substance is changed in the process.
Ex: Flammability, reactivity

Physical change - a change that does not change the identity of the substances involved.
Ex: dissolving, cutting, tearing, melting, freezing

Chemical change - A change that does change the identity of the substances involved. Bonds must be broken and new bonds formed.

Signs of chemical change: Bubbles (production of a gas), precipitate (production of a new solid), new color, new odor, change in temperature.

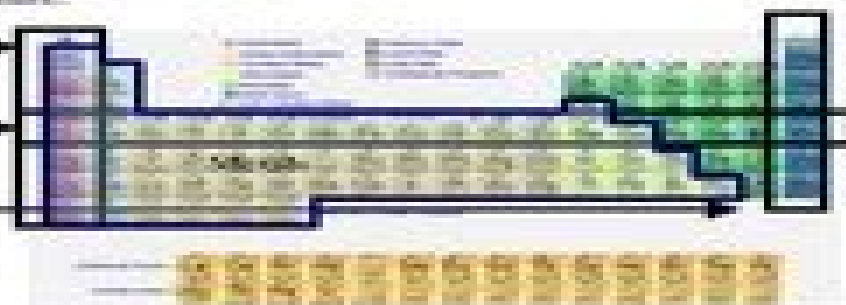
Ex: Combustion, oxidation

Periodic Table

Groups: the columns of the periodic table, elements in the same group have similar characteristics

Periods: the rows of the periodic table, tell you how many orbitals or shells that an atom of a particular element have

Noble Gases - unreactive gases in group 18



symbol



Name

Atomic number: equal to the number of protons in atoms of that element

Atomic mass: weighted average mass of all the isotopes of an element

Mass number: Atomic number rounded to the nearest whole number, number of protons + number of neutrons

Metals: The elements left of the jagged line. They are usually good conductors, are malleable, are ductile, and react to magnets. All are solids except mercury.

Nonmetals: The elements to the right of the jagged line. Their properties are opposite to those of metals.



Proton: Positively charged particle found in the nucleus

Neutron: Neutral particle found in the nucleus

Electrons: Negatively charged particles found orbiting the nucleus

Glynnis Lundberg

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