

- Directions:**  
 1. An element may be assigned a positive or negative charge depending on whether it is a metal or nonmetal.  
 2. An atom with a positive charge is called a cation.  
 3. An atom with a negative charge is called an anion.  
 4. The number of protons in an atom is the atomic number.

# Ion Practice Set

1. Write the symbol for the ion that has the same number of protons as the element but a different number of electrons.  
 2. Write the symbol for the ion that has the same number of protons as the element but a different number of electrons.

Complete the following table, using the periodic table in the back of your book.

	Element Name	Ion Symbol	Number of Protons	Number of Electrons	Number of Electrons Lost or Gained	Charge
1	Sodium	$\text{Na}^+$	11	10	lost one	positive
2	Chlorine	$\text{Cl}^-$	17	18	gained one	negative
3	Calcium	$\text{Ca}^{2+}$	20	18	lost two	positive
4	Fluorine	$\text{F}^-$	9	10	gained one	negative
5	Magnesium	$\text{Mg}^{2+}$	12	10	lost two	positive
6	Oxygen	$\text{O}^{2-}$	8	10	gained two	negative
7	Hydrogen	$\text{H}^+$	1	0	lost one	positive
8	Lithium	$\text{Li}^+$	3	2	lost one	positive
9	Boron	$\text{B}^{3+}$	5	2	lost three	positive
10	Carbon	$\text{C}^{4-}$	6	10	gained four	negative
11	Nitrogen	$\text{N}^{3-}$	7	10	gained three	negative
12	Aluminum	$\text{Al}^{3+}$	13	10	lost three	positive
13	Sulfur	$\text{S}^{2-}$	16	18	gained two	negative
14	Phosphorus	$\text{P}^{3-}$	15	18	gained three	negative
15	Selenium	$\text{Se}^{2-}$	34	36	gained two	negative
16	Bromine	$\text{Br}^-$	35	36	gained one	negative
17	Krypton	$\text{Kr}^{0}$	36	36	lost none	neutral
18	Xenon	$\text{Xe}^{0}$	54	54	lost none	neutral
19	Radium	$\text{Ra}^{2+}$	88	86	lost two	positive
20	Francium	$\text{Fr}^+$	87	86	lost one	positive

# Ion And Isotope Practice Set Answer Key

**Scott C. Dulebohn**



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