

Name: \_\_\_\_\_ Date: \_\_\_\_\_ Block: \_\_\_\_\_

## Ionic bonding practice

How many valence electrons do each of the following elements have?

- 1) carbon \_\_\_\_\_
- 2) selenium \_\_\_\_\_
- 3) xenon \_\_\_\_\_
- 4) potassium \_\_\_\_\_

Which of the following ions are likely to be formed? Circle one:

- |             |     |    |               |     |    |
|-------------|-----|----|---------------|-----|----|
| 5) $N^{-5}$ | yes | no | 8) $Al^{-2}$  | yes | no |
| 6) $He^{+}$ | yes | no | 9) $P^{-3}$   | yes | no |
| 7) $F^{-1}$ | yes | no | 10) $Mg^{-2}$ | yes | no |

Write the formula which result from combining the following ions:

- 11)  $K^{+1}$  and  $Br^{-}$  \_\_\_\_\_
- 12)  $Li^{+1}$  and  $O^{-2}$  \_\_\_\_\_
- 13)  $Ca^{+2}$  and  $Cl^{-1}$  \_\_\_\_\_
- 14)  $Al^{+3}$  and  $S^{-2}$  \_\_\_\_\_
- 15)  $Fe^{+2}$  and  $S^{-2}$  \_\_\_\_\_
- 16)  $Na^{+1}$  and  $NO_3^{-1}$  \_\_\_\_\_
- 17)  $Na^{+1}$  and  $SO_4^{-2}$  \_\_\_\_\_
- 18)  $NH_4^{+1}$  and  $CO_3^{-2}$  \_\_\_\_\_
- 19)  $Fe^{+3}$  and  $CrO_4^{-2}$  \_\_\_\_\_
- 20)  $Al^{+3}$  and  $PO_4^{-3}$  \_\_\_\_\_

# Ionic Bonding Practice Answer Key

**B Lingard**



## **Ionic Bonding Practice Answer Key:**

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