

Sodium becomes a cation, and chlorine becomes an anion.

Activity A: Ions	Get the Gizmo ready: <ul style="list-style-type: none"> Click Reset. Check that sodium and chlorine are still selected. 	
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Introduction: Some of the particles that make up atoms have an electrical charge. Electrons are negatively charged, while protons are positively charged. Particles with opposite charges (+ and -) attract, while particles with the same charge (+ and + or - and -) repel.

Question: What happens when atoms gain or lose electrons?

- Count:** Electrons move around the nucleus of atoms in specific shells, shown by the rings around the atoms in the Gizmo. The first ring holds two electrons, and the second holds eight. (Electrons in the inner rings are not shown; you can assume these rings are full.)

- Observe the sodium and chlorine atoms. Assuming that the inner rings are full of electrons, how many electrons are there total in each atom?

Sodium: 11 electrons

Chlorine: 17 electrons

- Each atom is neutrally charged, which means that each atom has the same number of protons and electrons. Based on this, how many protons are in each atom?

Sodium: 11 protons

Chlorine: 17 protons

- Observe:** Most atoms are stable with a configuration of eight valence electrons. This is known as the **octet rule**. How many valence electrons does each atom have?

Sodium: 1 valence electron

Chlorine: 7 valence electrons

- Form a bond:** Each electron has a charge of $1-$, and each proton has a charge of $1+$. You can calculate the charge of an atom by subtracting the number of electrons from the number of protons. Move an electron from the sodium to the chlorine atom.

- What are the charges of each atom now? Sodium: +1 Chlorine: -1

Turn on **Show charge** to check. These charged atoms are called **ions**.

- Is each ion stable? Explain.

Yes, each ion is stable. Chlorine only has 7 valence electrons, so gaining 1 more would lead to a stable full octet. Sodium on the other hand has 1 valence electron, by losing this electron it has just a regular shell of 8 electrons, also becoming a stable octet.

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Ionic Bonds Gizmo Explore Learning Answers

Hongru Du



Ionic Bonds Gizmo Explore Learning Answers:

The book delves into Ionic Bonds Gizmo Explore Learning Answers. Ionic Bonds Gizmo Explore Learning Answers is a vital topic that must be grasped by everyone, ranging from students and scholars to the general public. This book will furnish comprehensive and in-depth insights into Ionic Bonds Gizmo Explore Learning Answers, encompassing both the fundamentals and more intricate discussions.

1. This book is structured into several chapters, namely:
 - Chapter 1: Introduction to Ionic Bonds Gizmo Explore Learning Answers
 - Chapter 2: Essential Elements of Ionic Bonds Gizmo Explore Learning Answers
 - Chapter 3: Ionic Bonds Gizmo Explore Learning Answers in Everyday Life
 - Chapter 4: Ionic Bonds Gizmo Explore Learning Answers in Specific Contexts
 - Chapter 5: Conclusion
 2. In chapter 1, the author will provide an overview of Ionic Bonds Gizmo Explore Learning Answers. This chapter will explore what Ionic Bonds Gizmo Explore Learning Answers is, why Ionic Bonds Gizmo Explore Learning Answers is vital, and how to effectively learn about Ionic Bonds Gizmo Explore Learning Answers.
 3. In chapter 2, this book will delve into the foundational concepts of Ionic Bonds Gizmo Explore Learning Answers. This chapter will elucidate the essential principles that must be understood to grasp Ionic Bonds Gizmo Explore Learning Answers in its entirety.
 4. In chapter 3, the author will examine the practical applications of Ionic Bonds Gizmo Explore Learning Answers in daily life. This chapter will showcase real-world examples of how Ionic Bonds Gizmo Explore Learning Answers can be effectively utilized in everyday scenarios.
 5. In chapter 4, the author will scrutinize the relevance of Ionic Bonds Gizmo Explore Learning Answers in specific contexts. This chapter will explore how Ionic Bonds Gizmo Explore Learning Answers is applied in specialized fields, such as education, business, and technology.
 6. In chapter 5, this book will draw a conclusion about Ionic Bonds Gizmo Explore Learning Answers. The final chapter will summarize the key points that have been discussed throughout the book.
- This book is crafted in an easy-to-understand language and is complemented by engaging illustrations. It is highly recommended for anyone seeking to gain a comprehensive understanding of Ionic Bonds Gizmo Explore Learning Answers.

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