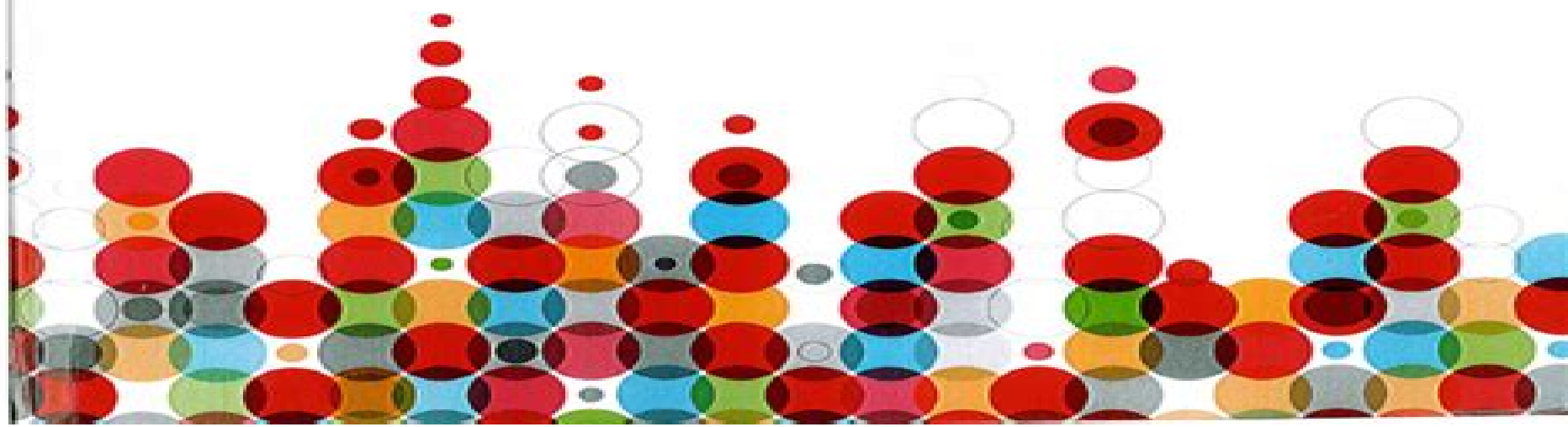


# Manual de Preparación para el Examen CISA

26° edición



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 $2\text{ }3\text{Ca(OH)}_2 + 1\text{ }Al_2\text{SO}_4 \rightarrow 4\text{ }3\text{CaSO}_4 + 2\text{ }Al(OH)_3$  type of reaction double displacement  
 $3\text{ }3\text{Mg} + 1\text{ }Fe_2O_3 \rightarrow 2\text{ }Fe + 3\text{ }MgO$  type of reaction single displacement  
 $4\text{ }1\text{ }C_2H_4 + 3\text{ }O_2 \rightarrow 2\text{ }CO_2 + 2\text{ }H_2O$

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 $NaOH + KNO_3 \rightarrow NaNO_3 + KOH$   
 $CH_2O + CO_2 + H_2 \rightarrow 2\text{ }H_2O + 2\text{ }Zn$   
 $2\text{ }Fe + 6\text{ }NaBr \rightarrow 2\text{ }FeBr_3 + 6\text{ }Na$   
 $3\text{ }Mg + 2\text{ }Fe_2O_3 \rightarrow 4\text{ }Fe + 3\text{ }MgO$   
 $4\text{ }2\text{ }NH_4OH + 4\text{ }HBr \rightarrow 4\text{ }H_2O + 4\text{ }NH_4Br$   
 $PbO + 2\text{ }Na_2CO_3 \rightarrow 2\text{ }Na_2O + CO_2 + C$   
 $4\text{ }H_2 + 8\text{ }FeO + 4\text{ }CO_2 \rightarrow 2\text{ }4\text{ }H_2O + 4\text{ }Fe + 4\text{ }CO$   
 $NaOH + H_2O \rightarrow NaCl$

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