# PRACTICE PAPER 🔯

#### SECTION-A (MULTIPLE CHOICE QUESTIONS)

- Read the following statements and choose the correct option.
  - (i) Amines can be distinguished by Hinsberg's reagent.
  - (ii) The three types of amines react similarly with Hinsberg's reagent and hence can be distinguished from each other.
  - (a) (i) is true and (ii) is true.
  - (i) is true and (ii) is false.
  - (c) (i) is false and (ii) is true.
  - (d) (i) is false and (ii) is false.
- 2. One mole of an organic compound A with the formula C,H,O reacts completely with two moles of HI to form X and Y. When Y is boiled with aqueous alkali, it forms Z. Z answers the iodoform test. The compound A is
  - (a) propan-2-ol
- (b) propan-1-ol
- (4d) methoxyethane. (c) ethoxyethane
- 3c 0.30 g of an organic compound containing C. H and O on combustion yields 0.44 g CO, and 0.18 g H.O. If one mol of compound weighs 60, then molecular formula of the compound is
  - (a) C<sub>2</sub>H<sub>2</sub>O

JOHN C.H.O.

(c) CH<sub>2</sub>O

- (d) C<sub>2</sub>H<sub>2</sub>O
- 1 The correct order of hybridisation of the central atom in the following species NH<sub>3</sub>. [PtCl<sub>4</sub>]<sup>2-</sup>, PCl<sub>4</sub> and BCl<sub>v</sub> is
  - (a) dsp2, dsp2, sp2, sp3, dsp2, dsp2, sp3d, sp2
  - (c) dsp<sup>2</sup>, sp<sup>2</sup>, sp<sup>3</sup>, dsp<sup>3</sup>(d) dsp<sup>2</sup>, sp<sup>3</sup>, sp<sup>2</sup>, dsp<sup>3</sup>
- Arrange the following compounds in increasing . 9 Consider the following statements : order of oxidation number.

#### MnCl<sub>2</sub>, MnO<sub>2</sub>, Mn(OH)<sub>3</sub>, KMnO<sub>4</sub>

- (a) MnCl<sub>2</sub> < MnO<sub>3</sub> < Mn(OH)<sub>3</sub> < KMnO<sub>4</sub>
- (b) MnO<sub>2</sub> < MnCl<sub>2</sub> < KMnO<sub>4</sub> < Mn(OH)<sub>3</sub>
- (c) Mn(OH)<sub>1</sub> < MnCl<sub>2</sub> < MnO<sub>2</sub> < KMnO<sub>4</sub>
- $MnCl_2 < Mn(OH)_3 < MnO_2 < KMnO_4$

- Elements X, Y and Z have atomic numbers 19. 37 and 55 respectively. Which of the following statements is true about them?
  - (a) Their ionisation potential would increase with increasing atomic number.
  - (b) Y would have an ionisation potential between those of X and Z.
  - (c) Z would have the highest ionisation potential.
  - (d) Y would have the highest ionisation potential.
- Statement-1: Bromobenzene upon reaction with Br<sub>2</sub>/Fe gives 1, 4-dibromobenzene as the major product.

Statement-2: In bromobenzene, the inductive effect of the bromo group is more dominant than the mesomeric effect in directing the incoming electrophile.

- (a) Statement-1 is true; statement-2 is true; statement-2. is a correct explanation for statement-1.
- (b) Statement-1 is true; statement-2 is true; statement-2 is not a correct explanation for statement-1.
- Statement-1 is true, statement-2 is false.
  - (d) Statement-1 is false, statement-2 is true.
- What would be the wavelength and name of series respectively for the emission transition for H-atom if it starts from the orbit having radius 1.3225 nm and ends at 211.6 pm?
  - لام) 434 nm, Balmer
- (b) 434 pm. Paschen
- (c) 545 pm, Pfund
- (d) 600 nm, Lyman
- - CH<sub>3</sub>OCH<sub>2</sub> is more stable than CH<sub>3</sub>CH<sub>2</sub>.
  - (II) Me<sub>3</sub>CH is more stable than CH<sub>3</sub>CH<sub>3</sub>CH<sub>3</sub>.
  - (III)  $CH_1 = CH CH_2$ , is more stable than CH<sub>3</sub>CH<sub>3</sub>CH<sub>3</sub>.
  - (IV) CH2 = CH is more stable than CH2CH2.

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