

- Directions:**
 1. An element may be assigned a positive or negative charge depending on whether it is a metal or nonmetal.
 2. An atom with a positive charge is called a cation.
 3. An atom with a negative charge is called an anion.
 4. The number of protons in an atom is the atomic number.

Ion Practice Set

1. Write the symbol for the ion that has the same number of protons as the element but a different number of electrons.
 2. Write the symbol for the ion that has the same number of protons as the element but a different number of electrons.

Complete the following table, using the periodic table in the back of your book.

	Element Name	Ion Symbol	Number of Protons	Number of Electrons	Number of Electrons Lost or Gained	Charge
1	Sodium	Na^+	11	10	lost one	positive
2	Chlorine	Cl^-	17	18	gained one	negative
3	Calcium	Ca^{2+}	20	18	lost two	positive
4	Oxygen	O^{2-}	8	10	gained two	negative
5	Aluminum	Al^{3+}	13	10	lost three	positive
6	Sulfur	S^{2-}	16	18	gained two	negative
7	Hydrogen	H^+	1	0	lost one	positive
8	Lithium	Li^+	3	2	lost one	positive
9	Fluorine	F^-	9	10	gained one	negative
10	Magnesium	Mg^{2+}	12	10	lost two	positive
11	Strontium	Sr^{2+}	38	36	lost two	positive
12	Barium	Ba^{2+}	56	54	lost two	positive
13	Strontium	Sr^{2+}	38	36	lost two	positive
14	Calcium	Ca^{2+}	20	18	lost two	positive
15	Sodium	Na^+	11	10	lost one	positive
16	Chlorine	Cl^-	17	18	gained one	negative

Ion Practice Set Answer Key

Shasha Hu

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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Table of Contents Ion Practice Set Answer Key

1. Understanding the eBook Ion Practice Set Answer Key
 - The Rise of Digital Reading Ion Practice Set Answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Ion Practice Set Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Ion Practice Set Answer Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Ion Practice Set Answer Key
 - Personalized Recommendations
 - Ion Practice Set Answer Key User Reviews and Ratings
 - Ion Practice Set Answer Key and Bestseller Lists
5. Accessing Ion Practice Set Answer Key Free and Paid eBooks
 - Ion Practice Set Answer Key Public Domain eBooks
 - Ion Practice Set Answer Key eBook Subscription Services
 - Ion Practice Set Answer Key Budget-Friendly Options

6. Navigating Ion Practice Set Answer Key eBook Formats
 - ePub, PDF, MOBI, and More
 - Ion Practice Set Answer Key Compatibility with Devices
 - Ion Practice Set Answer Key Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Ion Practice Set Answer Key
 - Highlighting and Note-Taking Ion Practice Set Answer Key
 - Interactive Elements Ion Practice Set Answer Key
8. Staying Engaged with Ion Practice Set Answer Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Ion Practice Set Answer Key
9. Balancing eBooks and Physical Books Ion Practice Set Answer Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Ion Practice Set Answer Key
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Ion Practice Set Answer Key
 - Setting Reading Goals Ion Practice Set Answer Key
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Ion Practice Set Answer Key
 - Fact-Checking eBook Content of Ion Practice Set Answer Key
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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